Novel Rhodium-Catalyzed Reaction of Thiazolidine Derivatives with **Carbodiimides**

Hai-Bing Zhou, Chune Dong, and Howard Alper*[a]

Abstract: A new, simple, and regioselective synthesis of thiazolidinimine derivatives based on the rhodium-catalyzed reaction of readily available thiazolidines with carbodiimides is described. This methodology provides direct access to a large variety of thiazolidinimine derivatives, possibly via a novel regiospecific insertion of carbodiimides into one of two ring carbon–nitrogen bonds, as well as a metal-catalyzed imine elimination process.

Introduction

The synthesis of five-membered heterocyclic rings by metalcatalyzed ring-expansion reactions of heterocyclic compounds has attracted considerable attention in recent years, because of the potential biological activity of some of the products.[1] The usual heterocyclic substrates for this type of reaction with heterocumulenes are small rings, such as oxiranes^[2] and aziridines;^[3] the reaction results in the formation of 1,3-oxazolidine and imidazolidine derivatives in good to excellent isolated yields. This reaction usually occurs when a substrate having π -electrons (e.g., phenyl) or a vinyl substituent is located at the 2-position of the strained ring system, but not with simple alkylheterocycles. One of us recently reported the first example of the ring expansion of thiiranes catalyzed by palladium to form thiazolidinimine derivatives.^[4] Some thiazolidine and thiazolidinimine moieties are found in molecules possessing biological activity.[5] For a variety of reasons including concerns about the incon-

[a] Dr. H.-B. Zhou, Dr. C. Dong, Prof. Dr. H. Alper Centre for Catalysis Research and Innovation Department of Chemistry, University of Ottawa, 10 Marie Curie Ottawa, Ontario, K1N 6N5 (Canada) Fax: (+1) 613-562-5271 E-mail: howard.alper@uottawa.ca

Supporting information for this article is available on the WWW under http://www.chemeurj.org/ or from the author: Selected X-ray data compound 3d.

Keywords: carbodiimides heterocycles · P ligands · rhodium · thiazolidine

venient access to thiiranes, and the benefits of having alternative procedures available for metal-catalyzed synthetic transformations, we have been developing new methods for the formation of functionalized thiazolidine derivatives from economical and readily available sources. Because of the supposed poisoning of the catalyst by the organic-sulfur re- $\arctan t$, $[6, 7]$ there are few examples of ring-expansion reactions catalyzed by transition-metal complexes.[8] There are no examples, to our knowledge, of the cyclization of a heterocycle containing two different heteroatoms with carbodiimides. The question arises as to what degree of selectivity of ring-opened cycloaddition occurs into rings containing two heteroatoms. In particular, the regioselectivity of the ring expansion reaction (insertion into carbon-nitrogen versus carbon-sulfur bonds of an N,S-containing heterocycle) is a matter of considerable interest. Herein, we describe the first rhodium-catalyzed rearrangement and cyclization reaction of thiazolidines with carbodiimides for the formation of thiazolidinimine derivatives. Moreover, this methodology uses simple thiazolidines as substrates, which could be readily prepared by alkylation of commercial available thiazolidines. The cyclization occurred under relatively mild reaction conditions (i.e., $90-130$ °C and 5 psi N₂). While the anticipated cyclization likely occurs, the reaction proceeds in a novel manner, affording thiazolidinimines in high yield. An unusual imine elimination step may be part of the overall process. The scope of the cyclization was successfully extended to simple unactivated thiazolidines and to functionalized thiazolidinimines.

Results and Discussion

The palladium-catalyzed reaction of 2-vinylpyrrolidine with heterocumulenes results in the formation of seven-mem-

6058 **External Constant On 2004 Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim DOI: 10.1002/chem.200400543** Chem. Eur. J. 2004, 10, 6058 – 6065

bered ring heterocycles.^[9,10] We anticipated that thiazolidines would also afford seven-membered ring heterocycles on reaction with heterocumulenes. Our initial attempts to realize the cyclization of thiazolidine 1a with aryl isocyanates failed, with starting materials recovered and a complex mixture of unknown compounds formed. When a carbodiimide was used as the reactant, for example, in the reaction of 3-[(ethoxycarbonyl)methyl]thiazolidine $(1a)$ with bis(pchlorophenyl)carbodiimide $(2d)$, in the presence of catalytic amounts of palladium acetate and dppf at 130° C for 3 d, the thiazolidinimine $3d$ was isolated in 22% yield (Table 1, PPh₃, dppp or dpppentane, affords thiazolidinimine 2d in up to 8% yield, along with the recovery of some of the starting material. However, increased loading of the catalyst did not improve the yield at all (Table 1, entries 1, 3 and 4). When $[PdCl₂(PhCN)₂]$ or $[PdCl₂(MeCN)₂]$ was used as the catalyst, only traces of the product was observed (Table 1, entries 5 and 6). In comparison, use of a rhodium complex and a phosphine ligand always afforded the desired product in relatively higher yield.

To optimize the present catalytic reaction, various phosphine ligands were systematically tested for the reaction of

Table 1. Reaction of 3-[(ethoxycarbonyl)methyl]-thiazolidine (1a) with bis(p-chlorophenyl)carbodiimide 2d catalyzed by a metal complex.[a]

Entry	Catalyst [mol%]	Conv. $[%]^{[b]}$	3d, Yields $\lceil\% \rceil^{ c }$
	$[Pd(OAc)_2]/PPh_3 (15/60)$	10	θ
2	$[Pd(OAc)2]$ /dppf (15/30)	31	22
3	[Pd(OAc) ₂]/dppp (15/30)	11	8
4	$[Pd(OAc)2]$ /dpppentane (15/30)	10	6
5	$[PdCl_2(PhCN)_2]$ (20)	8	≤ 5
6	$[PdCl2(MeCN)2]$ (20)		\lt 3
7	[$Rh(cod)Cl2/dppf (10/30)$	65	59
8	$[Rh(cod)Cl]_2/PPh_3 (10/60)$	58	50
9	$[Rh(cod)Cl]$ ₂ /dpppentane (10/30)	38	30
10	[Rh(cod)Cl] ₂ /dppp $(10/30)$	42	38
11	[$Rh(cod)Cl2/dppe$ (10/30)	31	28
12	$[Rh(cod)Cl]_2/P(tBu)$ ₃ (10/60)	<10	Ω
13	$[Rh(cod)Cl]_2/P(O)Ph_3 (10/60)$	<10	Ω
14	[$Rh(cod)Cl$], (10)	< 10	Ω

[a] The catalyst was dissolved in dry toluene, followed by addition of 1a and 2d, the mixture was transferred in a glass autoclave by syringe and stirred under 5 psi at 130 °C. [b] The conversion was determined by GC, using biphenyl as an internal standard, or was calculated based on the crude ¹H NMR of the mixture. [c] Isolated yield after preparative TLC and based on the thiazolidine 1a used.

entry 2). Dual-site functionalization occurred in this transformation, that is, formation of the thiazolidinimine moiety by rearrangement and cyclization of the thiazolidine ring and amide group formation by the amidation of the ester substituent in the reactant thiazolidine 1a.

Identifying features in the ${}^{1}H$ NMR spectra characteristic of 3d include the singlet for the methylene proton between the sulfur and nitrogen atoms, and the signals for the ester ethyl group disappear, while the singlet due to the acetamide is shifted downfield by approximately 1 ppm. The 13 C NMR spectrum displays a signal for the carbon of the imine of the thiazolidinimine at 162 ppm. Molecular ion peaks consistent with the structure are observed in the mass spectrum. The structure of thiazolidinimine 3d was also unambiguously established by X-ray determination (Figure 1).^[11] We then investigated optimization of the reaction of 1a with 2d by using different reaction conditions.

We found that treatment of the thiazolidine $1a$ with bis(p chlorophenyl)carbodiimide $(2c)$ in the presence of a catalytic amount of $[Pd(OAc)_2]$ and phosphine ligands such as 1a with 2d. The bidentate phosphine ligand dppf shows the highest catalytic activity. For example, reaction of 1a with $2d$ in the presence of 10 mol% $[Rh(cod)Cl]$ ₂ and 30 mol\% of dppf gave $3a$ in 59% yield and 65% conversion at 130° C for 3 d (Table 1, entry 7). In addition, we used $PPh₃$ as the added ligand, which has similar basicity to dppf, in the reaction of 1a with $2d$. Use of $10 \text{ mol } \%$ $[Rh(cod)Cl]_2$ and 60 mol% of PPh₃ gave $3d$ in 58% conversion and 50% yield respectively (Table 1, entry 8). Use of the more basic bidentate phosphine ligands dpppentane, dppp or dppe for the rhodium catalyzed reaction afforded 3 d in modest yields, that is, 30, 38, and 28% yields, respectively (Table 1, entries 9, 10 and 11). No products were observed by using a trialkylphosphine such as tri-n-butylphosphine or tri-

phenylphosphine oxide (Table 1, entries 12 and 13). It should be noted that simply heating $1a$ with $2d$ in the presence of a catalytic quantity of $[Rh(cod)Cl]_2$, but in the ab-

FULL PAPER **Herefore and Contain the Contract Container** H. Alper et al.

sence of a phosphine ligand, gave only starting material and some decomposition (Table 1, entry 14). Based on the reactions conditions above, we concluded that the catalytic system consisting of $[Rh(cod)Cl]_2$ and dppf is preferred for this reaction. Having established the reaction conditions affording the desired product 3 d, we next investigated the reaction of thiazolidines 1a–c with the carbodiimides $2a-f$.

The nature of the carbodiimide has a significant effect upon the reaction course. When diphenylcarbodiimide 1a was used, product 3a was obtained in 60% yield and 81% conversion in 72 h at 130° C (Table 2, entry 1). When carbodiimides containing electron-withdrawing substituents on the aromatic ring $(i.e., 2b-f)$ were used for the reaction with $1a$, the strength of the electron-withdrawing substituent on the aryl carbodiimide significantly affected the outcome of the reaction. For example, reaction of 1a with bis(chlorophenyl)carbodiimides 2b-d, proceeded smoothly and afforded thiazolidinimines 3b–d in moderate yields (40–59%) (Table 2, entries 2, 3 and 4). However, for bis(p-bromophenyl)carbodiimide $(2e)$, the reaction only occurred at the thiazolidine ring site, gave thiazolidinimine 3 e in 25% yield, without the amidation reaction at the ester position (Table 2, entry 5). Use of more reactive $bis(p\text{-nitro-}$ phenyl)carbodiimide $(2 f)$, which has an increased electron-withdrawing strength, in the reaction with $1a$, rather than the thiazolidinimine, gave the imidazolidinimine 4f in 42% yield and 52% conversion (Table 2, entry 6). It is conceivable that a thioketene elimination process occurred instead of imine elimination.

Table 2. (Continued)

[a] The catalyst was dissolved in dry toluene, followed by addition of 1 and 2, the mixture was transferred in a glass autoclave by syringe and stirred under 5 psi at specified temperature. [b] The conversion was determined by GC, by using biphenyl as an internal standard, or was calculated based on the crude ¹H NMR of the mixture. [c] Isolated yield after preparative TLC and based on the thiazolidine 1 used.

We next investigated the reaction of 3-butylthiazolidine 1**b** with carbodiimides 2**a–f** by using reaction conditions similar to those described above. Reaction of 1b with 2a-f was efficient and the products usually were isolated in reasonable yields. Reaction of 1b with diphenylcarbodiimide 2a only gave the thiazolidinimine 3g in 81% yield and 91% conversion (Table 2, entry 7). For reaction of bis(chlorophenyl)carbodiimides $2b-d$ with $1b$, the position of the chloro

substituent on the aryl carbodiimides has a significant effect on product formation. For example, reaction of 1b with bis(o-chlorophenyl)carbodiimide 2b gave the imidazolidinimine 4h as the major product in 76% yield and 98% conversion, accompanied by 15% of thiazolidinimine 3h (Table 2, entry 8). However, placing a chloro substituent at the meta- or para-position of the aryl carbodiimides (e.g., 2 c and $2d$), afforded both thiazolidinimines $3i$ and $3j$ in 36 and 54% yield, as well as imidazolidinimines $4i$ and $4j$ in 24 and 26% yield, respectively (Table 2, entries 9 and 10). Analogous to the reaction of 1**b** with 2**c** and 2**d**, bis(p -brobodiimide 2, possibly via a four-membered transition state **C**, may afford intermediate **D** or \mathbf{E} .^[13] Migratory insertion of aryl imido into the Rh–C bond of C would give D . Competitive insertion of carbene into the $Rh-N$ bond of C would give E. Subsequent imine elimination of D would give the intermediate F and alternatively, thioketene elimination of E would lead to the intermediate G. Insertion of carbene or aryl imido into the Rh-N or Rh-C bond of \bf{F} or \bf{G} may give

Scheme 1. Possible mechanism for the formation of 3.

mophenyl)carbodiimide (2e), in the reaction with 1_b , gave thiazolidinimine $3k$ and imidazolidinimine $4k$ in 41 and 33% yield, respectively (Table 2, entry 11). Reaction of 1b with bis(p-nitrophenyl)carbodiimide 2 f led to the exclusive formation of the imidazolidinimine 4l in 52% yield and 63% conversion (Table 2, entry 12).

When 3-(benzoylmethyl)thiazolidine $(1c)$ was treated with carbodiimide 2b and 2d under the same conditions, the desired products $3m$ and $3n$ were isolated in 52 and 34% yield, respectively (Table 2, entries 13 and 14). Some unidentified products were also formed in these reactions.

A possible mechanism for the reaction is illustrated in Scheme 1. The first step may involve the oxidative addition of the rhodium complex to the thiazolidine 1 would form the Rh^H complex **B.**^[12] Subsequent cycloaddition of B to the car-

H or I, followed by reductive elimination to form 3 or 4, and regenerate the catalyst. It seems reasonable to assume that the imino carbon of 3 or 4 is derived from the carbodiimide. To clarify this point, 13C enriched diphenylcarbodiimide 5 was prepared from benzoic-carboxy-13C acid. Treatment of this acid with sodium azide by using cyanuric chloride in the presence of 4-methylmorpholine afforded labeled benzoyl azide in 80% yield.^[14] The benzoyl azide underwent thermal rearrangement in refluxing benzene to form the labeled phenyl isocyanate.^[15] After completion of the reaction (GC) , the phosphine catalyst^[16] was added and the mixture was stirred at 65°C for 3 h. The solvent was removed and the residue was distilled under reduced pressure to afford pure labeled diphenylcarbodiimide 5. When 5 was treated with 3-butylthiazolidine 1b using conditions identical to those for the unlabeled reaction $\{[Rh(cod)Cl]_2, PhCH_3,$ dppf, 130° C, 3 d, the ¹³C-labeled thiazolidinimine 6 was obtained in 90% conversion and 78% yield [Eq. (2)].

The ¹³C NMR spectrum of 6 (in CDCl₃) clearly shows that the product contains ¹³C at the 2-position (δ 158.8 (¹³C=N)) (Figure 2), which is a very weak peak in 13 C NMR of 3g (Figure 3). The mass spectrum gave an intense molecular ion at m/e 235. Therefore, the ¹³C-labeling experiment demonstrated that the source of the new atom in 6 is the imido group of 5.

In conclusion, we have achieved a facile direct synthesis of thiazolidinimine derivatives from readily available alkylthiazolidines. The reported catalyst system is tolerant to a variety of thiazolidines and carbodiimides, making the procedure valuable for the synthesis of interesting heterocycles of potential pharmaceutical use.

Experimental Section

All reactions and manipulations of chemicals were carried out using standard Schlenk techniques under an atmosphere of argon. Alkylthiazolidines^[8] and carbodiimides^[16] were prepared according to the literature. Toluene was dried over Na prior to use. All NMR spectra were recorded using CDCl₃ as the solvent with reference to residual CHCl₃ (1 H at 7.24 ppm and 13 C at 77.0 ppm). Infrared spectra were recorded on a Fourier transform spectrometer and are reported in wavenumbers $(cm⁻¹)$.

General procedure for the rhodium-catalyzed cycloaddition reaction of alkylthiazolidines (1a-c) with carbodiimides 2: $[Rh(cod)Cl]_2$ (14.8 mg, 0.03 mmol, 10 mol% to 1) was weighed into a Schlenk tube under a stream of argon, and dry toluene (3 mL) was added. Dppf (49.86 mg, 30 mol%) was then added, followed by 1a, 1b, or 1c (0.3 mmol) and then carbodiimide 2 (0.45 mmol). The mixture was transferred in a glass

autoclave and stirred under 5 psi of $N₂$ at a given temperature. The progress of the reaction was monitored by GC and the crude product was purified by silica chromatography using hexane/ethyl acetate 10:1 to 1:1 to afford thiazolidinimine 3 or imidazolidinimine 4. Further purification was effected using preparative TLC.

N-Phenyl-2-(2-phenyliminothiazolidin-3-yl)-acetamide $(3a)$: 60% yield;

colorless oil; IR (neat): $\tilde{v} = 3290, 3056, 2937, 2867, 1672, 1612, 1587, 1546,$ 1496, 1442, 1294, 1248, 1190, 1142, 1026, 758, 694 cm⁻¹; ¹H NMR $(300 \text{ MHz}, \text{CDCI}_3): \delta = 3.21 \text{ (t, } J = 6.9 \text{ Hz}, 2H), 3.78 \text{ (t, } J = 6.9 \text{ Hz}, 2H),$ 4.26 (s, 2H), 6.97–7.52 (m, 10H), 9.46 (b, 1H); 13C NMR (75 MHz, CDCl₃): $\delta = 167.1, 161.1, 150.5, 137.9, 129.2, 129.0, 124.2, 123.9, 121.9,$ 119.7, 52.8, 52.1, 27.2; MS (70 eV, EI): m/z (%): 311 (10) $[M^+]$; EI-HRMS: m/z : calcd for $C_{17}H_{17}N_3OS$: 311.109249; found: 311.10776.

N-(2-Chlorphenyl)-2-[2-(2-chlorophenylimino)-thiazolidin-3-yl]-acet-

amide (3b): 43% yield; colorless oil; IR (neat): $\tilde{v} = 3367$, 3060, 2943, 2868, 1693, 1609, 1581, 1524, 1469, 1440, 1292, 1238, 1190, 1126, 1057, 1033, 924, 756, 729 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ = 3.29 (t, J =

Figure 2. 13C NMR spectrum of 6.

Figure 3. 13 C NMR spectrum of 3g.

6.9 Hz, 2H), 3.84 (t, J=6.9 Hz, 2H), 4.40 (s, 2H), 6.96–7.07 (m, 3H), 7.15–7.20 (m, 1H), 7.23–7.28 (m, 1H), 7.32–7.36 (m, 2H), 8.25 (dd, $J=$ 8.2 Hz, 1H), 9.10 (b, 1H); ¹³C NMR (75 MHz, CDCl₃): δ = 166.7, 162.3, 134.3, 130.5, 129.8, 129.2, 127.9, 127.5, 127.3, 125.4, 124.3, 123.5, 122.8, 52.6, 52.0, 27.5; MS (70 eV, EI): m/z (%): 379 (6) $[M^+]$, 381 (3) $[M^+ +2]$; EI-HRMS: m/z : calcd for C₁₇H₁₅N₃OSCl₂: 379.031309; found: 379.03269.

N-(3-Chlorphenyl)-2-[2-(3-chlorophenylimino)-thiazolidin-3-yl]-acet-

amide (3c): 42% yield; colorless oil; IR (neat): $\tilde{v} = 3285$, 3065, 2947, 2872, 1678, 1614, 1582, 1481, 1425, 1294, 1190, 1089, 1072, 995, 874, 844, 779, 692 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ = 3.24 (t, J = 7.0 Hz, 2H), 3.79 (t, J=7.0 Hz, 2H), 4.23 (s, 2H), 6.83–6.87 (m, 1H), 6.97–6.98 (m, 1H), 7.05–7.07 (m, 2H), 7.18–7.26 (m, 3H), 7.67–7.68 (m, 1H), 9.34 (b, 1H); ¹³C NMR (75 MHz, CDCl₃): δ = 166.9, 161.7, 151.7, 138.8, 134.7, 134.5, 130.2, 130.0, 124.3, 123.9, 122.2, 120.2, 119.8, 117.6, 52.6, 52.1, 27.3; MS (70 eV, EI): m/z (%): 379 (6) $[M^+]$, 381 (23) $[M^+ +2]$; EI-HRMS: m/z : calcd for C₁₇H₁₅N₃OSCl₂: 379.031309; found: 379.031408.

N-(4-Chlorphenyl)-2-[2-(4-chlorophenylimino)-thiazolidin-3-yl]-acet-

amide (3d): 59% yield; colorless needles; m.p. 143–145°C; IR (neat): \tilde{v} = 3273, 3053, 2956, 2869, 1674, 1614, 1583, 1539, 1489, 1400, 1327, 1301, 1240, 1190, 1143, 1089, 1010, 923, 831, 731, 704 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ = 3.23 (t, J = 7.0 Hz, 2H), 3.80 (t, J = 7.0 Hz, 2H), 4.28 (s, 2H), 6.88–6.91 (m, 2H), 7.23–7.27 (m, 4H), 7.38–7.44 (m, 2H), 9.47 (b, 1H); ¹³C NMR (75 MHz, CDCl₃): δ = 166.7, 162.0, 148.5, 136.4, 129.3, 129.2, 129.1, 129.0, 123.3, 120.8, 52.6, 52.4, 27.3; MS (70 eV, EI): m/z (%): 379 (4) $[M^+]$, 381 (6) $[M^+ +2]$; EI-HRMS: m/z : calcd for C₁₇H₁₅N₃OSCl₂: 379.031309;found: 379.03159.

[2-(4-Bromophenylimino)-thiazolidin-3-yl]-acetic acid ethyl ester (3 e): 25% yield; colorless oil; IR (neat): $\tilde{v} = 3057, 2979, 2929, 2871, 1745, 1614,$ 1577, 1485, 1442, 1417, 1373, 1296, 1238, 1203, 1097, 1070, 1010, 927, 879, 833, 711 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ = 1.28 (t, J = 7.1 Hz, 1H), 3.16 (t, $J=6.9$ Hz, 2H), 3.80 (t, $J=6.9$ Hz, 2H), 4.24 (q, $J=7.1$ Hz, 2H), 4.44 (br, 2H), 6.89 (d, J=7.9 Hz, 2H), 7.37 (d, J=7.9 Hz, 2H); ¹³C NMR $(75 \text{ MHz}, \text{CDCl}_3): \delta = 168.4, 161.2, 132.0, 131.5, 124.5, 120.2, 61.6, 48.3,$ 29.7, 27.5, 14.2; MS (70 eV, EI): m/z (%): 342 (100) $[M^+]$, 344 (98) $[M^+]$ +2]; EI-HRMS: m/z : calcd for $C_{13}H_{15}N_2O_2SBr$: 342.003791; found: 341.99644.

[3-(4-Nitrophenyl)-2-(4-nitrophenylimino)-imidazolidin-1-yl]-acetic acid ethyl ester (4 f): 42% yield; colorless oil; IR (neat): $\tilde{v} = 3379, 3114, 3080,$ 2981, 2904, 1741, 1634, 1595, 1573, 1504, 1435, 1323, 1280, 1201, 1101, 1024, 953, 912, 854, 752, 735, 692 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ =

1.21 (t, $J=7.1$ Hz, 3H), 3.72 (t, $J=7.7$ Hz, 2H), 3.90 (s, 2H), 4.07 (t, $J=$ 7.7 Hz, 2H), 4.14 (q, $J=7.1$ Hz, 2H), 6.77 (d, $J=8.8$ Hz, 2H), 7.41(d, $J=$ 9.1 Hz, 2H), 7.95 (d, $J=8.8$ Hz, 2H), 8.05 (d, $J=9.1$ Hz, 2H); ¹³C NMR $(75 \text{ MHz}, \text{CDCl}_3)$: $\delta = 168.6, 150.0, 146.0, 142.9, 141.6, 125.0, 124.5, 121.3,$ 120.2, 113.3, 61.7, 48.6, 47.5, 46.1, 14.1; MS (70 eV, EI): m/z (%): 413 (100) $[M^+]$; EI-HRMS: m/z : calcd for C₁₉H₁₉N₅O₆: 413.133509; found: 413.1327.

(3-Butyl-thiazolidin-2-ylidene)-phenyl-amine (3g): 81% yield; colorless oil; IR (neat): $\tilde{v} = 3053$, 2957, 2927, 2860, 1622, 1587, 1489, 1441, 1406, 1377, 1333, 1292, 1234, 1184, 1105, 1070, 1024, 926, 877, 767, 696 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): $\delta = 0.95$ (t, J = 7.3 Hz, 3H), 1.32–1.44 (m, 2H), 1.57–1.66 (m, 2H), 3.10 (t, J=6.9 Hz, 2H), 3.50 (t, J=7.3 Hz, 2H), 3.60 (t, J=6.9 Hz, 2H), 6.90–6.93 (m, 2H), 6.98–7.02 (m, 1H), 7.22–7.27 $(m, 2H)$; ¹³C NMR (75 MHz, CDCl₃): δ = 158.8, 135.3, 128.7, 122.9, 122.1, 50.9, 46.3, 29.3, 26.9, 20.1, 13.9; MS (70 eV, EI): m/z (%): 234 (70) $[M^+]$; EI-HRMS: m/z : calcd for C₁₃H₁₈N₂S: 234.11909; found: 234.11907.

(3-Butyl-thiazolidin-2-ylidene)-(2-chlorophenyl)-amine (3 h): 15% yield; colorless oil; IR (neat): $\tilde{v} = 3060, 2958, 2927, 2858, 1645, 1625, 1581, 1519,$ 1475, 1440, 1298, 1238, 1126, 1055, 1033, 752, 727, 667 cm⁻¹; ¹H NMR $(300 \text{ MHz}, \text{CDCl}_3)$: $\delta = 0.92$ (t, $J = 7.3 \text{ Hz}, 3 \text{ H}$), 1.33–1.42 (m, 2H), 1.54– 1.63 (m, 2H), 3.10 (t, J=6.9 Hz, 2H), 3.50 (t, J=7.3 Hz, 2H), 3.621 (t, $J=6.9$ Hz, 2H), 6.80–6.98 (m, 3H), 7.15 (m, 1H); ¹³C NMR (75 MHz, CDCl₃): δ = 159.3, 132.1, 131.5, 128.8, 128.0, 123.6, 120.3, 51.0, 46.3, 29.3, 27.0, 20.1, 13.9; MS (70 eV, EI): m/z (%): 268 (75) [M⁺], 270 (28) [M⁺ +2]; EI-HRMS: m/z : calcd for C₁₃H₁₇N₂SCl: 268.08012; found: 268.0821.

[1-Butyl-3-(2-chlorophenyl)-imidazolidin-2-ylidene]-(2-chlorophenyl)-

amine (4h): 76% yield; colorless oil; IR (neat): $\tilde{v} = 3059$, 2956, 2927, 2869, 1645, 1579, 1481, 1437, 1419, 1367, 1273, 1099, 1053, 1031, 933, 864, 756, 729, 698 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ = 0.96 (t, J = 7.4 Hz, 3H), 1.35–1.46 (m, 2H), 1.60–1.70 (m, 2H), 3.41 (t, J=6.5 Hz, 2H), 3.53 $(t, J=7.4 \text{ Hz}, 2H)$, 3.66 (b, 2H), 6.39-6.46 (m, 1H), 6.55-6.69 (m, 2H), 6.83–6.93 (m, 3H), 7.06–7.12 (m, 2H); ¹³C NMR (75 MHz, CDCl₃): δ = 152.1, 146.3, 139.2, 131.8, 129.4, 128.8, 128.4, 127.5, 127.0, 126.6, 126.0, 123.6, 121.0, 48.9, 45.7, 45.2, 29.2, 20.1, 14.0; MS (70 eV, EI): m/z (%): 361 (13) $[M^+]$, 363 (8) $[M^+ +2]$, 365 (6) $[M^+ +4]$; EI-HRMS: m/z : calcd for $C_{19}H_{21}N_3Cl_2$: 361.111245; found: 361.10829.

(3-Butyl-thiazolidin-2-ylidene)-(3-chlorophenyl)-amine (3i): 36% yield; colorless oil; IR (neat): $\tilde{v} = 3059, 2956, 2929, 2860, 1612, 1581, 1556, 1467,$ 1441, 1404, 1290, 1234, 1184, 1105, 1070, 995, 925, 906, 871, 844, 779, 717, 694 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ = 0.94 (t, J = 7.3 Hz, 3H), 1.31–

FULL PAPER H. Alper et al.

1.43 (m, 2H), 1.55–1.65 (m, 2H), 3.12 (t, $J=6.9$ Hz, 2H), 3.49 (t, $J=$ 7.3 Hz, 2H), 3.62 (t, $J=6.9$ Hz, 2H), 6.81 (d, $J=8.1$ Hz, 1H), 6.93–6.98 (m, 2H), 7.15 (t, J=7.9 Hz, 1H); ¹³C NMR (75 MHz, CDCl₃): δ = 159.5, 134.1, 131.5, 129.7, 122.9, 122.5, 120.5, 51.0, 46.3, 29.3, 26.9, 20.1, 13.9; MS (70 eV, EI): m/z (%): 268 (16) $[M^+]$, 270 (6) $[M^+ +2]$; EI-HRMS: m/z : calcd for C₁₃H₁₇N₂SCl: 269.08012 (M⁺+1); found: 269.08100.

[1-Butyl-3-(3-chlorophenyl)-imidazolidin-2-ylidene]-(3-chlorophenyl)-

amine (4i): 24% yield; colorless oil; IR (neat): $\tilde{v} = 3060, 2958, 2929, 2871,$ 1633, 1579, 1479, 1409, 1371, 1269, 1112, 1095, 993, 900, 871, 777, 686 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ = 0.88 (t, J = 7.3 Hz, 3H), 1.23– 1.28 (m, 2H), 1.50–1.61 (m, 2H), 3.36 (t, $J=7.3$ Hz, 2H), 3.57 (t, $J=$ 8.0 Hz, 2H), 3.87 (t, J=8.0 Hz, 2H), 6.77 (m, 3H), 6.89–6.92 (m, 2H), 7.03–7.05 (m, 3H); ¹³C NMR (75 MHz, CDCl₃): δ = 151.7, 144.3, 141.8, 134.2, 133.8, 129.5, 129.3, 124.7, 122.7, 122.1, 121.6, 121.0, 120.5, 48.7, 46.7, 45.2, 29.0, 19.9, 13.8;MS (70 eV, EI): m/z (%): 361 (29) [M ⁺], 363 (18) $[M^+ +2]$, 365 (8) $[M^+ +4]$; EI-HRMS: m/z : calcd for C₁₉H₂₁N₃Cl₂: 361.111245;found: 361.1076.

(3-Butyl-thiazolidin-2-ylidene)-(4-chlorophenyl)-amine (3 j): 54% yield; colorless oil; IR (neat): $\tilde{v} = 3032, 2956, 2929, 2860, 1616, 1583, 1487, 1441,$ 1408, 1296, 1232, 1184, 1091, 1009, 833, 715, 700 cm⁻¹; ¹H NMR $(300 \text{ MHz}, \text{CDCl}_3)$: $\delta = 0.94$ (t, $J = 7.4 \text{ Hz}, 3 \text{ H}$), 1.30–1.40 (m, 2H), 1.55– 1.65 (m, 2H), 3.11 (t, $J=6.9$ Hz, 2H), 3.49 (t, $J=7.4$ Hz, 2H), 3.60 (t, $J=$ 6.9 Hz, 2H), 6.85 (d, J=6.7 Hz, 2H), 7.19 (dd, J=6.7 Hz, 2H); 13C NMR $(75 \text{ MHz}, \text{CDC}1_3): \delta = 159.5, 150.5, 128.8, 128.1, 123.6, 51.0, 46.3, 29.3,$ 26.9, 20.1, 13.9; MS (70 eV, EI): m/z (%): 268 (6) $[M^+]$, 270 (3) $[M^+]$ +2]; EI-HRMS: m/z : calcd for C₁₃H₁₇N₂Cl: 269.08012; found: 269.0849 [M⁺ $+H$].

[1-Butyl-3-(4-chlorophenyl)-imidazolidin-2-ylidene]-(4-chlorophenyl)-

amine (4j): 26% yield; colorless oil; IR (neat): $\tilde{v} = 3033$, 2958, 2929, 2871, 1633, 1583, 1492, 1417, 1404, 1371, 1271, 1114, 1091, 1012, 827, 700 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ = 0.87 (t, J = 7.3 Hz, 3H), 1.23– 1.31 (m, 2H), 1.49–1.59 (m, 2H), 3.37 (t, $J=7.3$ Hz, 2H), 3.57 (t, $J=$ 7.8 Hz, 2H), 3.86 (t, J=7.8 Hz, 2H), 6.74 (t, J=8.3 Hz, 2H), 6.98 (m, 2H), 7.03 (m, 4H); ¹³C NMR (75 MHz, CDCl₃): δ = 152.3, 143.2, 137.6, 134.9, 128.8, 128.4, 123.4, 118.3, 49.2, 46.8, 45.2, 29.0, 19.8, 13.8; MS (70 eV, EI): m/z (%): 361 (26) $[M^+]$, 363 (17) $[M^+ +2]$, 365 (8) $[M^+ +4]$; EI-HRMS: m/z : calcd for C₁₉H₂₁N₃Cl₂: 361.111245; found: 361.1151.

(3-Butyl-thiazolidin-2-ylidene)-(4-bromophenyl)-amine (3 k): 41% yield; colorless oil; IR (neat): $\tilde{v} = 3055$, 2956, 2927, 2860, 1614, 1577, 1483, 1438, 1409, 1332, 1294, 1234, 1184, 1105, 1068, 1004, 924, 879, 831, 731, 692 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ = 0.93 (t, J = 7.3 Hz, 3H), 1.31– 1.40 (m, 2H), 1.53–1.65 (m, 2H), 3.12 (t, $J=6.9$ Hz, 2H), 3.48 (t, $J=$ 7.3 Hz, 2H), 3.61 (t, J=6.9 Hz, 2H), 6.87 (d, J=6.8 Hz, 2H), 7.38 (dd, $J=6.8$ Hz, 2H); ¹³C NMR (75 MHz, CDCl₃): $\delta=154.1, 132.0, 131.4,$ 128.3, 124.8, 52.1, 47.4, 29.7, 27.6, 20.1, 14.0; MS (70 eV, EI): m/z (%): 312 (76) $[M^+]$, 314 (77) $[M^+]$; EI-HRMS: m/z : calcd for $C_{13}H_{17}N_2SBr: 312.029613$; found: 312.02800.

[1-Butyl-3-(4-bromophenyl)-imidazolidin-2-ylidene]-(4-bromophenyl)-

amine (4k): 33% yield; colorless oil; IR (neat): $\tilde{v} = 3024$, 2956, 2927, 2869, 1633, 1575, 1479, 1417, 1404, 1375, 1321, 1271, 1095, 1070, 1010, 927, 823, 734 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ = 0.88 (t, J = 7.4 Hz, 3H), 1.25–1.31 (m, 2H), 1.48–1.59 (m, 2H), 3.18 (t, J=7.4 Hz, 2H), 3.49 $(t, J=7.9 \text{ Hz}, 2H), 3.76 (t, J=7.9 \text{ Hz}, 2H), 6.61 (d, J = 7.9 \text{ Hz}, 2H),$ 6.97–7.09 (m, 4H), 7.23 (d, J=7.9 Hz, 2H); ¹³C NMR (75 MHz, CDCl₃): δ = 151.8, 139.1, 131.8, 131.6, 131.4, 129.5, 124.1, 123.9, 118.1, 49.2, 46.9, 45.3, 29.0, 19.8, 13.8;MS (70 eV, EI): m/z (%): 449 (29) [M ⁺], 451 (58) $[M^+ +2]$, 453 (26) $[M^+ +4]$; EI-HRMS: m/z : calcd for C₁₉H₂₁N₃Br₂: 449.010231; found: 449.01198.

[1-Butyl-3-(4-nitrophenyl)-imidazolidin-2-ylidene]-(4-nitrophenyl)-amine (4l): 52% yield; colorless oil; IR (neat): $\tilde{v} = 3082, 2958, 2929, 2872, 1633,$ 1595, 1574, 1500, 1479, 1429, 1321, 1271, 1172, 1139, 1109, 1049, 852, 752, 694 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ = 0.81 (t, J = 7.2 Hz, 3H), 1.14– 1.25 (m, 2H), 1.42–1.49 (m, 2H), 3.21 (t, $J=7.3$ Hz, 2H), 3.68 (t, $J=$ 7.2 Hz, 2H), 4.05 (t, J=7.3 Hz, 2H), 6.87 (d, J=8.4 Hz, 2H), 7.44 (d, J= 8.4 Hz, 2H), 7.96–8.17 (m, 4H); ¹³C NMR (75 MHz, CDCl₃): δ = 150.2, 146.1, 142.6, 126.3, 125.0, 124.5, 121.1, 119.6, 113.3, 60.4, 46.8, 45.0, 28.7, 19.7, 13.6; MS (70 eV, EI): m/z (%): 383 (28) [M⁺]; EI-HRMS: m/z : calcd for $C_{19}H_{21}N_5O_4$: 383.159331; found: 383.15924.

2-[2-(2-Chlorophenylimino)-thiazolidin-3-yl]-1-phenyl-ethanone (3m): 52% yield; colorless oil; IR (neat): $\tilde{v} = 3059, 2956, 2924, 2872, 1697, 1624,$

1582, 1467, 1438, 1350, 1292, 1222, 1180, 1057, 1031, 912, 831, 756, 729, 688 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ = 3.25 (t, J = 6.9 Hz, 2H), 3.80 $(t, J=6.9 \text{ Hz}, 2\text{ H}), 5.02 \text{ (s, 2H)}, 6.94-6.98 \text{ (m, 2H)}, 7.15 \text{ (t, } J=7.5 \text{ Hz},$ 1H), 7.33 (d, J=8.0 Hz, 1H), 7.45 (t, J=7.8 Hz, 2H), 7.58 (t, J=7.3 Hz, 1H), 8.03–8.06 (m, 2H); ¹³C NMR (75 MHz, CDCl₃): δ = 194.6, 159.5, 135.1, 133.6, 129.7, 128.9, 128.3, 127.2, 125.5, 121.6, 52.7, 51.6, 27.3; MS (70 eV, EI): m/z (%): 330 (43) $[M^+]$, 332 (15) $[M^+]$ +2]; EI-HRMS: m/z : calcd for $C_{17}H_{15}N_3Cl_2OSCl$: 330.059384; found: 330.06056.

2-[2-(4-Chlorophenylimino)-thiazolidin-3-yl]-1-phenyl-ethanone (3 n): 34% yield; colorless needles; m.p. 95–97 °C; IR (neat): $\tilde{v} = 3059$, 2957, 2923, 2874, 1697, 1620, 1583, 1489, 1439, 1419, 1400, 1294, 1222, 1180, 1144, 1089, 1012, 912, 877, 833, 754, 690 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ = 3.27 (t, J = 6.9 Hz, 2H), 3.79 (t, J = 6.9 Hz, 2H), 5.07 (s, 2H), 6.89–6.94 (m, 2H), 7.19 (d, J=7.9 Hz, 1H), 7.42–7.49 (m, 2H), 7.60–7.61 (m, 2H), 8.01 (d, J=7.3 Hz, 2H); ¹³C NMR (75 MHz, CDCl₃): δ =194.4, 161.2, 135.1, 133.4, 129.6, 127.9, 124.2, 125.5, 121.6, 61.7, 48.3, 29.6; MS (70 eV, EI): m/z (%): 330 (50) $[M^+]$, 332 (21) $[M^+ +2]$; EI-HRMS: m/z : calcd for $C_{17}H_{15}N_3Cl_2OSCl$: 330.059384; found: 330.05772.

Procedure for the rhodium-catalyzed reaction of 3-butylthiazolidine 1 b and 13 C diphenylcarbodiimide 5—(3-Butyl-thiazolidin-2-ylidene)[2- 13 C]phenyl-amine (6): The reaction procedure was the same as that described in the general procedure. The isolated yield of pure $\bf{6}$ was 78%; colorless oil; IR (neat): $\tilde{v} = 3056$, 2956, 2854, 1624, 1575, 1493, 1438, 1227, 1176, 1068, 1022, 760 cm⁻¹; ¹H NMR (500 MHz, CDCl₃): δ = 0.97 (t, J = 7.3 Hz, 3H), 1.36–1.43 (m, 2H), 1.59–1.66 (m, 2H), 3.14 (t, J=6.9 Hz, 2H), 3.50 $(t, J=7.3 \text{ Hz}, 2H)$, 3.60 $(t, J=6.9 \text{ Hz}, 2H)$, 6.92–6.93 $(m, 2H)$, 7.00–7.04 (m, 1H), 7.24–7.28 (m, 2H); ¹³C NMR (125 MHz, CDCl₃): δ = 158.8, 135.3, 128.7, 122.9, 122.1, 50.9, 46.2, 29.3, 26.8, 20.1, 13.9;MS (70 eV, EI): m/z (%): 235 (60) [M⁺]; EI-HRMS: m/z : calcd for C₁₃H₁₈N₂S: 235.1226; found: 235.1218.

Acknowledgement

We are grateful to the Natural Sciences and Engineering Research Council of Canada for support of this research. We thank Dr. X. Ou-Yang for performing the X-ray crystallographic analysis of $3d$. We appreciate the donation of $[Rh(cod)Cl]_2$ by Johnson-Mathey Corporation.

- [1] a) B. El Ali, H. Alper, in Transition Metal Catalyzed Reactions (Eds.: S.-I. Murahashi, S. G. Davies), Blackwell Science, Malden, MA, 1999, p. 261, and references therein; b) A. R. Katritzky, C. W. Rees, in Comprehensive Heterocyclic Chemistry, Pergamon Press, Oxford, 1984, Part 1 and 4B.
- [2] a) C. Larksarp, H. Alper, J. Org. Chem. 1998, 63, 6229; b) C. Larksarp, H. Alper, J. Am. Chem. Soc. 1997, 119, 3709; c) G. P. Speranza, W. J. Peppel, *J. Org. Chem.* **1958**, 23, 1922; d) J. E. Herweh, T. A. Foglia, D. Swern, *J. Org. Chem.* **1968**, 33, 4029; e) J. E. Herweh, W. J. Kauffman, Tetrahedron Lett. 1971, 809; f) A. Baba, M. Fujiwara, H. Matsuda, Tetrahedron Lett. 1986, 27, 77; g) M. Fujiwara, A. Baba, H. Matsuda, J. Heterocycl. Chem. 1988, 25, 1351; h) I. Shibata, A. Baba, H. Iwasaki, H. Matsuda, J. Org. Chem. 1986, 51, 2177; i) A. Baba, K. Seki, H. Matsuda, J. Heterocycl. Chem. 1990, 27, 1925; j) B. M. Trost, A. R. Sudhakar, J. Am. Chem. Soc. 1987, 109, 3792; k) C. Qian, D. Zhu, Synlett 1994, 129; l) M. Brunner, L. Mußmann, D. Vogt, Synlett 1994, 69.
- [3] a) D. C. D. Butler, G. A. Inman, H. Alper, J. Org. Chem. 2000, 65, 5887;b) J. O. Baeg, C. Bensimon, H. Alper, J. Am. Chem. Soc. 1995, 117, 4700; c) H. Maas, C. Bensimon, H. Alper, J. Org. Chem. 1998, 63, 17;d) J. O. Baeg, H. Alper, J. Org. Chem. 1992, 57, 157;e) J. O. Baeg, H. Alper, J. Am. Chem. Soc. 1994, 116, 1220; f) U.K. Nadir, N. Basu, *J. Org. Chem.* **1995**, 60, 1458; g) J. Sepulveda-Arques, T. Armero-Alarte, A. Acero-Alarcón, E. Zaballos-Garcia, B. Y. Solesio, J. E. Carrera, Tetrahedron 1996, 52, 2097.
- [4] C. Larksarp, O. Sellier, H. Alper, J. Org. Chem. 2001, 66, 3502.
- [5] a) T. Suzuki, H. Nagaoka, Y. Kondo, T. Takahashi, M. Takeuchi, Chem. Pharm. Bull. 1998, 46, 1468; b) C. A. Gandolfi, R. Di Domenico, S. Spinelli, L. Gallico, L. Fiocchi, E. M. Lotto, A. Borghi, C. Dalla Rosa, S. Tognella, J. Med. Chem. 1995, 38, 508;c) K. D. Klika,

L. Janovec, G. Suchár, P. Kristian, R. Sillanpää, K. Pihlaja, Eur. J. Org. Chem. 2002, 1248.

- [6] a) R. Hughes, "Catalyst deactivation by poisoning", in Deactivation of Catalyst, Academic Press, Orlando, 1984, p. 81, Chapter 5; b) B. Delmon, G. F. Froment, Catalyst Deactivation: Proceedings of the International Symposium, Antwerp, Oct 13–15, 1980, Elsevier, New York, 1980.
- [7] a) J. Oudar, "The Role of Sulfur in Catalyst Deactivation", in Catalyst Deactivation (Eds.: E. E. Peterson, A. J. Bell), Marcel Dekker, New York, 1987, p. 149;b) C. H. Bartholomew, P. K. Agrawal, J. R. Katzer, "Sulfur Poisoning of Metals", in Advances in Catalysis, Vol. 31 (Eds.: D. D. Eley, H. Pines, P. B. Weisz), Academic Press, New York, 1982, p. 135.
- [8] a) K. Khumtaveeporn, H. Alper, J. Am. Chem. Soc. 1994, 116, 5662; b) M. D. Wang, S. Calet, H. Alper, J. Org. Chem. 1989, 54, 20.
- [9] H. B. Zhou, H. Alper, J. Org. Chem. 2003, 68, 3439.
- [10] H. B. Zhou, H. Alper, *Tetrahedron* 2004, 60, 73.
- [11] CCDC-246783 $(3d)$ contains the supplementary crystallographic data. These data can be obtained free of charge via www.ccdc.cam. ac.uk/conts/retrieving.html (or from the Cambridge Crystallographic

Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: (+44) 1223-336-033; or deposit@ccdc.cam.uk).

- [12] One reviewer suggested that the Rh moiety would insert in the $C-S$ bond rather than the C-N bond. We proposed that the Rh moiety inserts in $C-N$ bond because our previous study (ref. [8a]) has shown, in the rhodium(i) catalyzed ring expansion carbonylation of thiazolidines, a six-membered-ring heterocycle was obtained with exclusive carbon monoxide insertion into the C2-N bond and no insertion into carbon-sulfur bond. These results indicated that the active rhodium intermediate was generated by oxidative addition of rhodium (i) into the C2-N bond of thiazolidine.
- [13] See ref. [8a] for rhodium catalyzed ketene elimination process; b) for an example of the transformation of an imido rhodium complex, see F. Ragaini, S. Cenini, Organometallics 1994, 13, 1178.
- [14] B. P. Bandgar, S. S. Pandit, *Tetrahedron Lett.* **2002**, 43, 3413.
- [15] M. A. Masson, J. R. Dormoy, J. Labelled Compd. Radiopharm. 1979, 16, 785.
- [16] T.W. Campbell, J.J. Monagle, V.S. Fold, J. Am. Chem. Soc. 1962, 84, 3673.

Received: May 31, 2004 Published online: October 28, 2004